

4. Hausübung, **Statistische Physik**

abzugeben am Donnerstag, 10.11.2011

Aufgabe H7 *Gibbs' paradox* (5 Punkte)

Recall that the partition function for one free particle of mass M in a box of volume N is

$$Z_1 = n_Q(\tau)V$$

where

$$n_Q(\tau) = \left(\frac{M\tau}{2\pi\hbar^2} \right)^{\frac{3}{2}}.$$

- a. Consider N independent such particles, and assume that they are *distinguishable*. Show that the entropy is

$$\sigma(N, V) = N \ln(n_Q(\tau)V) + \frac{3}{2}N.$$

- b. Consider two identical boxes of volume V at temperature τ , each containing a gas of N distinguishable particles, and compute the total entropy for this system. Show that the difference between the total entropy for the two separate boxes and the total entropy for a single box of volume $2V$ containing $2N$ particles is

$$\Delta\sigma = -2N \ln 2.$$

This means that the entropy can be increased or decreased by a macroscopic amount just by inserting a barrier in the center of the large box or removing it, which contradicts the principles of thermodynamics.

- c. Do the same calculation when the particles are indistinguishable and show that, for $N \gg 1$,

$$\Delta\sigma \simeq 0.$$

(more problems on the next page..)

Aufgabe H8 *Gas in a potential* (4 Punkte)

Consider a classical particle with phase space coordinates (x, p) and Hamiltonian

$$H(x, p) = t(p) + v(x)$$

where t and v are real functions. Also we assume that the position x is restricted to a volume V . For the purpose of computing entropies or partition functions, we will count one state per phase space volume h . Let Z_1 be the canonical partition function of that system at temperature τ in the case where $v(x) = 0$.

- a. Allowing for more than one particle, and assuming the particles do not interact and are *indistinguishable*, show that the grand-canonical partition function at temperature τ and with fugacity $\lambda = e^{\mu/\tau}$ is

$$\mathcal{Z} = e^{\frac{\lambda Z_1}{V} \int e^{-v(x)/\tau} dx}.$$

- b. Show that the average number of particles is then

$$\langle N \rangle = \frac{Z_1}{V} \int e^{(\mu - v(x))/\tau} dx.$$

We will use this in the next problem.

Aufgabe H9 *Centrifuge* (3 Punkte)

Consider a circular cylinder of radius R and length L , rotating about its symmetry axis with angular velocity ω and containing an ideal gas with particles of mass M . We assume that the system is in thermal equilibrium at temperature τ , that the gas is at rest in a reference frame rotating with the cylinder, and that the particles' velocities are small enough that we can ignore all the Coriolis forces.

If $n(0)$ is the expected particle density on the axis of rotation, show that the expected particle density at a distance r is

$$n(r) = n(0) e^{\frac{1}{2} M \omega^2 r^2 / \tau}.$$

Hint: Consider a small volume of gas a distance r from the axis. Observe that it is at thermal and diffusive equilibrium with the rest of the gas and subject to a centrifugal potential $v(r)$. Then use the result of problem H8, recalling that the partition function for one particle of mass M in a box of volume V and at temperature τ has the form

$$Z_1 = \left(\frac{\tau}{\alpha} \right)^{\frac{3}{2}} V,$$

for some fixed energy $\alpha > 0$.